Subdivided Module Catalogue
for the Subject
Chemistry
as a Bachelor’s with 1 major
with the degree "Bachelor of Science"
(180 ECTS credits)

Examination regulations version: 2017
Responsible: Faculty of Chemistry and Pharmacy
Course of Studies - Contents and Objectives
The degree subject Chemistry is offered by the Faculty of Chemistry and Pharmacy of the JMU as a fundamental course of studies with the degree "Bachelor of Science" (B.Sc.) within the framework of a consecutive Bachelor and Master study model.

The Bachelor’s degree programme aims to impart the specialised knowledge and skills necessary for the transition to professional practice or a subsequent Master’s degree programme. These include the theoretical foundations of inorganic, organic, physical, theoretical and bioorganic chemistry, as well as the accompanying practical laboratory techniques. The course places emphasis on the acquisition of basic knowledge and training in scientific observation and practical laboratory skills, with a progressively increasing focus on independence in a laboratory setting. This intensive training should equip students with the confidence and skills needed for independent work in a range of application-oriented areas of industry, commerce or management, or provide the basis for advanced scientific training as part of a Master’s degree.
Abbreviations used

Course types: E = field trip, K = colloquium, O = conversatorium, P = placement/lab course, R = project, S = seminar, T = tutorial, Ü = exercise, V = lecture

Term: SS = summer semester, WS = winter semester

Methods of grading: NUM = numerical grade, B/NB = (not) successfully completed

Regulations: (L)ASPO = general academic and examination regulations (for teaching-degree programmes), FSB = subject-specific provisions, SFB = list of modules

Other: A = thesis, LV = course(s), PL = assessment(s), TN = participants, VL = prerequisite(s)

Conventions

Unless otherwise stated, courses and assessments will be held in German, assessments will be offered every semester and modules are not creditable for bonus.

Notes

Should there be the option to choose between several methods of assessment, the lecturer will agree with the module coordinator on the method of assessment to be used in the current semester by two weeks after the start of the course at the latest and will communicate this in the customary manner.

Should the module comprise more than one graded assessment, all assessments will be equally weighted, unless otherwise stated below.

Should the assessment comprise several individual assessments, successful completion of the module will require successful completion of all individual assessments.

In accordance with

the general regulations governing the degree subject described in this module catalogue:

ASPO2015

associated official publications (FSB (subject-specific provisions)/SFB (list of modules)):

9-Aug-2017 (2017-48)

This module handbook seeks to render, as accurately as possible, the data that is of statutory relevance according to the examination regulations of the degree subject. However, only the FSB (subject-specific provisions) and SFB (list of modules) in their officially published versions shall be legally binding. In the case of doubt, the provisions on, in particular, module assessments specified in the FSB/SFB shall prevail.
## The subject is divided into

<table>
<thead>
<tr>
<th>Abbreviation</th>
<th>Module title</th>
<th>ECTS credits</th>
<th>Method of grading</th>
<th>page</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Compulsory Courses (150 ECTS credits)</strong></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td><strong>Subfield General and Inorganic Chemistry (47 ECTS credits)</strong></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>08-AC1-152-m01</td>
<td>Principles of Inorganic Chemistry</td>
<td>8</td>
<td>NUM</td>
<td>8</td>
</tr>
<tr>
<td>08-ACP1-152-m01</td>
<td>Inorganic Chemistry 1 (lab)</td>
<td>10</td>
<td>B/NB</td>
<td>21</td>
</tr>
<tr>
<td>08-AS1-152-m01</td>
<td>Inorganic Chemistry of the Elements</td>
<td>6</td>
<td>NUM</td>
<td>13</td>
</tr>
<tr>
<td>08-ANP-152-m01</td>
<td>Analytical Chemistry (lab)</td>
<td>6</td>
<td>B/NB</td>
<td>20</td>
</tr>
<tr>
<td>08-ACP2-172-m01</td>
<td>Inorganic Chemistry 2 (lab)</td>
<td>5</td>
<td>B/NB</td>
<td>37</td>
</tr>
<tr>
<td>08-AC-FSE-152-m01</td>
<td>Solid State Chemistry, Spectroscopic Methods, Organoelement Chemistry</td>
<td>12</td>
<td>NUM</td>
<td>22</td>
</tr>
<tr>
<td><strong>Subfield Organic Chemistry (40 ECTS credits)</strong></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>08-OC1-152-m01</td>
<td>Organic Chemistry 1</td>
<td>5</td>
<td>NUM</td>
<td>14</td>
</tr>
<tr>
<td>08-OC2-152-m01</td>
<td>Organic Chemistry 2 and analytical methods in organic chemistry</td>
<td>9</td>
<td>NUM</td>
<td>16</td>
</tr>
<tr>
<td>08-OCP1-172-m01</td>
<td>Organic Chemistry - lab 1</td>
<td>8</td>
<td>B/NB</td>
<td>38</td>
</tr>
<tr>
<td>08-OCP2-152-m01</td>
<td>Organic Chemistry - advanced laboratory course for students of chemistry</td>
<td>5</td>
<td>B/NB</td>
<td>18</td>
</tr>
<tr>
<td>08-OC3+4-152-m01</td>
<td>Organic Chemistry 3 &amp; 4</td>
<td>13</td>
<td>NUM</td>
<td>23</td>
</tr>
<tr>
<td><strong>Subfield Physical and Theoretical Chemistry (40 ECTS credits)</strong></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>08-PC-QMS-152-m01</td>
<td>Principles of quantum mechanics and spectroscopy</td>
<td>10</td>
<td>NUM</td>
<td>24</td>
</tr>
<tr>
<td>08-PC-TKE-152-m01</td>
<td>Thermodynamics, Kinetics, Electrochemistry</td>
<td>9</td>
<td>NUM</td>
<td>15</td>
</tr>
<tr>
<td>08-PCP-152-m01</td>
<td>Physical Chemistry (lab)</td>
<td>9</td>
<td>B/NB</td>
<td>25</td>
</tr>
<tr>
<td>08-TC-152-m01</td>
<td>Quantum Chemistry</td>
<td>3</td>
<td>NUM</td>
<td>10</td>
</tr>
<tr>
<td>08-PC-SBL-152-m01</td>
<td>Symmetry, chemical bonding and light</td>
<td>9</td>
<td>NUM</td>
<td>19</td>
</tr>
<tr>
<td><strong>Subfield Basics of Natural Sciences (23 ECTS credits)</strong></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>08-BC1-152-m01</td>
<td>Biochemistry 1</td>
<td>5</td>
<td>NUM</td>
<td>9</td>
</tr>
<tr>
<td>10-M-MCH-172-m01</td>
<td>Mathematics for students in Chemistry and Biochemistry</td>
<td>5</td>
<td>NUM</td>
<td>36</td>
</tr>
<tr>
<td>11-EFNF-152-m01</td>
<td>Introduction to Physics for Students of other Disciplines</td>
<td>7</td>
<td>NUM</td>
<td>6</td>
</tr>
<tr>
<td>11-PFNF-152-m01</td>
<td>Laboratory Course Physics for Students of other Disciplines</td>
<td>3</td>
<td>B/NB</td>
<td>7</td>
</tr>
<tr>
<td>03-TR-152-m01</td>
<td>Toxicology and legal studies</td>
<td>3</td>
<td>NUM</td>
<td>11</td>
</tr>
<tr>
<td><strong>Key Skills Area (20 ECTS credits)</strong></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td><strong>General Key Skills (5 ECTS credits)</strong></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Students may select any of the modules offered as part of the pool of general transferable skills (ASQ) of JMU.</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td><strong>Subject-specific Key Skills (15 ECTS credits)</strong></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>08-VP-152-m01</td>
<td>Advanced laboratory course</td>
<td>5</td>
<td>B/NB</td>
<td>26</td>
</tr>
<tr>
<td><strong>Subject-specific Key Skills, Compulsory Electives (10 ECTS credits)</strong></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>08-BC2-152-m01</td>
<td>Biochemistry 2</td>
<td>5</td>
<td>NUM</td>
<td>12</td>
</tr>
<tr>
<td>08-BCP-152-m01</td>
<td>Practical course of Biochemistry</td>
<td>5</td>
<td>B/NB</td>
<td>17</td>
</tr>
<tr>
<td>08-PS3-152-m01</td>
<td>Applied Spectroscopy 3</td>
<td>5</td>
<td>NUM</td>
<td>27</td>
</tr>
<tr>
<td>08-PKC-152-m01</td>
<td>Programming and numerical methods</td>
<td>5</td>
<td>B/NB</td>
<td>28</td>
</tr>
<tr>
<td>08-OP-152-m01</td>
<td>Advanced chemical practical course</td>
<td>5</td>
<td>B/NB</td>
<td>29</td>
</tr>
<tr>
<td><strong>Thesis (10 ECTS credits)</strong></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>08-BA-152-m01</td>
<td>Bachelor Thesis</td>
<td>10</td>
<td>NUM</td>
<td>30</td>
</tr>
</tbody>
</table>
### Compulsory Electives, Appendix DA (170 ECTS credits)

<table>
<thead>
<tr>
<th>Code</th>
<th>Title</th>
<th>Credits</th>
<th>Exam Type</th>
<th>ECTS</th>
</tr>
</thead>
<tbody>
<tr>
<td>08-ANP-152-m01</td>
<td>Analytical Chemistry (lab)</td>
<td>6</td>
<td>B/NB</td>
<td>20</td>
</tr>
<tr>
<td>08-AC1-152-m01</td>
<td>Principles of Inorganic Chemistry</td>
<td>8</td>
<td>NUM</td>
<td>8</td>
</tr>
<tr>
<td>08-AS1-152-m01</td>
<td>Inorganic Chemistry of the Elements</td>
<td>6</td>
<td>NUM</td>
<td>13</td>
</tr>
<tr>
<td>08-ACP1-152-m01</td>
<td>Inorganic Chemistry 1 (lab)</td>
<td>10</td>
<td>B/NB</td>
<td>21</td>
</tr>
<tr>
<td>08-AC-FS-DA-152-m01</td>
<td>Solid State Chemistry, Spectroscopic Methods (DD)</td>
<td>5</td>
<td>NUM</td>
<td>31</td>
</tr>
</tbody>
</table>

### Subfield General and Inorganic Chemistry (35 ECTS credits)

<table>
<thead>
<tr>
<th>Code</th>
<th>Title</th>
<th>Credits</th>
<th>Exam Type</th>
<th>ECTS</th>
</tr>
</thead>
<tbody>
<tr>
<td>08-ANP-152-m01</td>
<td>Analytical Chemistry (lab)</td>
<td>6</td>
<td>B/NB</td>
<td>20</td>
</tr>
<tr>
<td>08-AC1-152-m01</td>
<td>Principles of Inorganic Chemistry</td>
<td>8</td>
<td>NUM</td>
<td>8</td>
</tr>
<tr>
<td>08-AS1-152-m01</td>
<td>Inorganic Chemistry of the Elements</td>
<td>6</td>
<td>NUM</td>
<td>13</td>
</tr>
<tr>
<td>08-ACP1-152-m01</td>
<td>Inorganic Chemistry 1 (lab)</td>
<td>10</td>
<td>B/NB</td>
<td>21</td>
</tr>
</tbody>
</table>

### Subfield Organic Chemistry (28 ECTS credits)

<table>
<thead>
<tr>
<th>Code</th>
<th>Title</th>
<th>Credits</th>
<th>Exam Type</th>
<th>ECTS</th>
</tr>
</thead>
<tbody>
<tr>
<td>08-OC2-152-m01</td>
<td>Organic Chemistry 2 and analytical methods in organic chemistry</td>
<td>9</td>
<td>NUM</td>
<td>16</td>
</tr>
<tr>
<td>08-OC1-152-m01</td>
<td>Organic Chemistry 1</td>
<td>5</td>
<td>NUM</td>
<td>14</td>
</tr>
<tr>
<td>08-OCP1-172-m01</td>
<td>Organic Chemistry - lab 1</td>
<td>8</td>
<td>B/NB</td>
<td>38</td>
</tr>
<tr>
<td>08-OC-OC3-DA-152-m01</td>
<td>Organic Chemistry 3 (DD)</td>
<td>6</td>
<td>NUM</td>
<td>32</td>
</tr>
</tbody>
</table>

### Subfield Physical and Theoretical Chemistry (37 ECTS credits)

<table>
<thead>
<tr>
<th>Code</th>
<th>Title</th>
<th>Credits</th>
<th>Exam Type</th>
<th>ECTS</th>
</tr>
</thead>
<tbody>
<tr>
<td>08-TC-152-m01</td>
<td>Quantum Chemistry</td>
<td>3</td>
<td>NUM</td>
<td>10</td>
</tr>
<tr>
<td>08-PC-TKE-152-m01</td>
<td>Thermodynamics, Kinetics, Electrochemistry</td>
<td>9</td>
<td>NUM</td>
<td>15</td>
</tr>
<tr>
<td>08-PC-QMS-152-m01</td>
<td>Principles of quantum mechanics and spectroscopy</td>
<td>10</td>
<td>NUM</td>
<td>24</td>
</tr>
<tr>
<td>08-PCP-152-m01</td>
<td>Physical Chemistry (lab)</td>
<td>9</td>
<td>B/NB</td>
<td>25</td>
</tr>
<tr>
<td>08-PC-SBL-DA-152-m01</td>
<td>Symmetry, chemical bonding and light (DD)</td>
<td>6</td>
<td>NUM</td>
<td>33</td>
</tr>
</tbody>
</table>

### Subfield Basics of Natural Sciences (20 ECTS credits)

<table>
<thead>
<tr>
<th>Code</th>
<th>Title</th>
<th>Credits</th>
<th>Exam Type</th>
<th>ECTS</th>
</tr>
</thead>
<tbody>
<tr>
<td>11-EFNF-152-m01</td>
<td>Introduction to Physics for Students of other Disciplines</td>
<td>7</td>
<td>NUM</td>
<td>6</td>
</tr>
<tr>
<td>11-PFNF-152-m01</td>
<td>Laboratory Course Physics for Students of other Disciplines</td>
<td>3</td>
<td>B/NB</td>
<td>7</td>
</tr>
<tr>
<td>08-BC1-152-m01</td>
<td>Biochemistry 1</td>
<td>5</td>
<td>NUM</td>
<td>9</td>
</tr>
<tr>
<td>10-M-MCH-172-m01</td>
<td>Mathematics for students in Chemistry and Biochemistry</td>
<td>5</td>
<td>NUM</td>
<td>36</td>
</tr>
</tbody>
</table>

### Subfield Competences from foreign university (50 ECTS credits)

<table>
<thead>
<tr>
<th>Code</th>
<th>Title</th>
<th>Credits</th>
<th>Exam Type</th>
<th>ECTS</th>
</tr>
</thead>
<tbody>
<tr>
<td>08-VPUB1-152-m01</td>
<td>Qualifications - Partner University 1</td>
<td>25</td>
<td>B/NB</td>
<td>34</td>
</tr>
<tr>
<td>08-VPUB2-152-m01</td>
<td>Qualifications - Partner University 2</td>
<td>25</td>
<td>NUM</td>
<td>35</td>
</tr>
</tbody>
</table>

### Thesis (10 ECTS credits)

<table>
<thead>
<tr>
<th>Code</th>
<th>Title</th>
<th>Credits</th>
<th>Exam Type</th>
<th>ECTS</th>
</tr>
</thead>
<tbody>
<tr>
<td>08-BA-152-m01</td>
<td>Bachelor Thesis</td>
<td>10</td>
<td>NUM</td>
<td>30</td>
</tr>
<tr>
<td><strong>Module title</strong></td>
<td><strong>Abbreviation</strong></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>------------------</td>
<td>------------------</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Introduction to Physics for Students of other Disciplines</td>
<td>11-EFNF-152-m01</td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

**Module coordinator**
Managing Director of the Institute of Applied Physics

**Module offered by**
Faculty of Physics and Astronomy

<table>
<thead>
<tr>
<th><strong>ECTS</strong></th>
<th><strong>Method of grading</strong></th>
<th><strong>Only after succ. compl. of module(s)</strong></th>
</tr>
</thead>
<tbody>
<tr>
<td>7</td>
<td>numerical grade</td>
<td>--</td>
</tr>
</tbody>
</table>

**Duration**
2 semester

**Module level**
undergraduate

**Other prerequisites**
--

**Contents**
Mechanics, vibration theory, thermodynamics, optics, science of electricity, Atomic and Nuclear Physics.

**Intended learning outcomes**
The students have knowledge of the principles of Physics.

**Courses**
(type, number of weekly contact hours, language — if other than German)

V (4) + V (3)

**Method of assessment**
(type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)

written examination (60 to 120 minutes)

**Allocation of places**
--

**Additional information**
--

**Referred to in LPO I**
(examination regulations for teaching-degree programmes)

--
Module title | Laboratory Course Physics for Students of other Disciplines
Abbreviation | 11-PFNF-152-m01

Module coordinator | Managing Director of the Institute of Applied Physics
Module offered by | Faculty of Physics and Astronomy

ECTS | 3
Method of grading | Only after succ. compl. of module(s)
(not) successfully completed | --

Duration | 1 semester
Module level | undergraduate
Other prerequisites | --

Contents
Simple experiments in the fields of mechanics, vibration theory, thermodynamics, optics, X-rays, nuclear magnetic resonance, Atomic and Nuclear Physics, imaging methods.

Intended learning outcomes
The students have detected and understood physical contexts on the basis of the implementation of own experiments. They have a basic understanding of physical phenomena and know the basic ideas and ways of functioning of different measuring and imaging methods as well as their applications, especially in the field of Biomedicine.

Courses (type, number of weekly contact hours, language — if other than German)
P (4)

Method of assessment (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)
a) practical assignment with oral test (approx. 15 minutes, during experiments) and b) written examination (90 minutes).
Each experiment comprises preparation, performance and evaluation. Test as well as performance of experiments can each be repeated once.

Allocation of places
Only as part of pool of general transferable skills (ASQ): 10 places (lottery)

Additional information
--

Referred to in LPO I (examination regulations for teaching-degree programmes)
--
<table>
<thead>
<tr>
<th>Module title</th>
<th>Abbreviation</th>
</tr>
</thead>
<tbody>
<tr>
<td>Principles of Inorganic Chemistry</td>
<td>08-AC1-152-m01</td>
</tr>
</tbody>
</table>

**Module coordinator**
- lecturer of lecture "Experimentalchemie" (Experimental Chemistry)

**Module offered by**
- Institute of Inorganic Chemistry

<table>
<thead>
<tr>
<th>ECTS</th>
<th>Method of grading</th>
<th>Only after succ. compl. of module(s)</th>
</tr>
</thead>
<tbody>
<tr>
<td>8</td>
<td>numerical grade</td>
<td>--</td>
</tr>
</tbody>
</table>

**Duration**
- 1 semester

**Module level**
- undergraduate

**Other prerequisites**
- --

**Contents**
Basics of general and anorganic chemistry.

**Intended learning outcomes**
Kenntnis der Grundlagen der Allgemeinen und Anorganischen Chemie

**Courses** (type, number of weekly contact hours, language — if other than German)
- V (4) + V (2)

**Method of assessment** (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)

- a) written examination (approx. 90 to 180 minutes) or b) oral examination of one candidate each (20 to 30 minutes) or c) oral examination in groups of up to 3 candidates (approx. 15 minutes per candidate) or d) log (approx. 20 pages) or e) presentation (approx. 30 minutes)

**Language of assessment**: German and/or English

**Allocation of places**
- --

**Additional information**
- --

**Referred to in LPO I** (examination regulations for teaching-degree programmes)
- § 42 I Nr. 1 and § 22 II Nr. 1 h)
- § 62 I Nr. 1
Module title
Biochemistry 1

Abbreviation
08-BC1-152-m01

Module coordinator
holder of the Chair of Biochemistry

Module offered by
Chair of Biochemistry

ECTS
5

Method of grading
numerical grade

Only after succ. compl. of module(s)
--

Duration
1 semester

Module level
undergraduate

Other prerequisites
--

Contents
The module imparts the basic knowledge of biochemistry by lectures and tutorials. Main topics of the module Biochemistry 1 are particularly the biochemistry of proteins (amino acids, peptide bond, primary, secondary, tertiary and quaternary structure), catalytic strategies and enzyme kinetics, carbohydrate metabolism (glycolysis, gluconeogenesis, citric acid cycle, cellular respiration, photosynthesis), fatty acid metabolism (beta-oxidation, fatty acid synthesis), nucleotide metabolism, urea cycle and metabolism of amino acids. Additionally the module conveys basic knowledge about the structure of DNA and the basics of passing and transformation of genetic information (central dogma).

Intended learning outcomes
The student has basic knowledge in the covered subject areas of biochemistry. He/She is able to describe the basic biochemical processes in cellular systems.

Courses (type, number of weekly contact hours, language — if other than German)
V (2) + Ü (1)

Method of assessment (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)
written examination (approx. 60 to 90 minutes)

Allocation of places
--

Additional information
--

Referred to in LPO I (examination regulations for teaching-degree programmes)
§ 42 I Nr. 2
§ 62 I Nr. 2
Module title  | Abbreviation
---|---
Quantum Chemistry  | 08-TC-152-m01

Module coordinator  | Module offered by
Lecturer of lecture "Quantenchemie"  | Institute of Physical and Theoretical Chemistry

ECTS  | Method of grading  | Only after succ. compl. of module(s)
3  | Numerical grade  | --

Duration  | Module level  | Other prerequisites
1 semester  | Undergraduate  | --

Contents

German contents available but not translated yet.


Intended learning outcomes

German intended learning outcomes available but not translated yet.

Die Studierenden sind in der Lage, mit Hilfe grundlegender Konzepte und Modelle angeregte Zustände von Molekülen zu beschreiben.

Courses (type, number of weekly contact hours, language — if other than German)

V (2) + Ü (1)

Method of assessment (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)

a) written examination (approx. 90 to 180 minutes) or b) oral examination of one candidate each (20 to 30 minutes) or c) oral examination in groups of up to 3 candidates (approx. 15 minutes per candidate) or d) log (approx. 20 pages) or e) presentation (approx. 30 minutes)

Language of assessment: German and/or English creditable for bonus

Allocation of places

--

Additional information

--

Referred to in LPO I (examination regulations for teaching-degree programmes)

§ 22 II Nr. 1 h)
§ 22 II Nr. 2 f)
§ 22 II Nr. 3 f)
<table>
<thead>
<tr>
<th>Module title</th>
<th>Abbreviation</th>
</tr>
</thead>
<tbody>
<tr>
<td>Toxicology and legal studies</td>
<td>03-TR-152-m01</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Module coordinator</th>
<th>Module offered by</th>
</tr>
</thead>
<tbody>
<tr>
<td>lecturer of lecture &quot;Toxikologie und Rechtskunde&quot;</td>
<td>Faculty of Medicine</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>ECTS</th>
<th>Method of grading</th>
<th>Only after succ. compl. of module(s)</th>
</tr>
</thead>
<tbody>
<tr>
<td>3</td>
<td>numerical grade</td>
<td>--</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Duration</th>
<th>Module level</th>
<th>Other prerequisites</th>
</tr>
</thead>
<tbody>
<tr>
<td>1 semester</td>
<td>undergraduate</td>
<td>--</td>
</tr>
</tbody>
</table>

**Contents**
Basics of legal regulations for chemists (handling and transportation of hazardous materials), fundamentals of toxicology.

**Intended learning outcomes**
The students master the basics of legal regulations for chemists (handling and transport of hazardous substances) as well as the fundamentals of toxicology.

**Courses** (type, number of weekly contact hours, language — if other than German)
V (1) + V (1)

**Method of assessment** (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)
written examination (approx. 90 minutes)

**Allocation of places**
--

**Additional information**
--

**Referred to in LPO I** (examination regulations for teaching-degree programmes)
§ 22 II Nr. 1 h)
§ 22 II Nr. 2 f)
§ 22 II Nr. 3 f)
## Module Catalogue for the Subject Chemistry

Bachelor’s with 1 major, 180 ECTS credits

<table>
<thead>
<tr>
<th>Module title</th>
<th>Abbreviation</th>
</tr>
</thead>
<tbody>
<tr>
<td>Biochemistry 2</td>
<td>08-BC2-152-m01</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Module coordinator</th>
<th>Module offered by</th>
</tr>
</thead>
<tbody>
<tr>
<td>holder of the Chair of Biochemistry</td>
<td>Chair of Biochemistry</td>
</tr>
</tbody>
</table>

### ECTS, Method of grading, Only after succ. compl. of module(s)

<table>
<thead>
<tr>
<th>ECTS</th>
<th>Method of grading</th>
<th>Only after succ. compl. of module(s)</th>
</tr>
</thead>
<tbody>
<tr>
<td>5</td>
<td>numerical grade</td>
<td>--</td>
</tr>
</tbody>
</table>

### Duration, Module level, Other prerequisites

<table>
<thead>
<tr>
<th>Duration</th>
<th>Module level</th>
<th>Other prerequisites</th>
</tr>
</thead>
<tbody>
<tr>
<td>1 semester</td>
<td>undergraduate</td>
<td>--</td>
</tr>
</tbody>
</table>

## Contents

The module imparts the basic knowledge of biochemistry by lectures and in-depth tutorials. Main topics of the module Biochemistry 2 are particularly Replication, DNA repair, transcription, mRNA maturation, translation and regulation of translation, protein targeting, nuclear transport, and protein degradation. Additionally the module conveys basic knowledge about the cellular signal transduction.

### Intended learning outcomes

The student has basic knowledge in the covered subject areas of biochemistry. He/She is able to describe the basic biochemical processes in cellular systems.

### Courses (type, number of weekly contact hours, language — if other than German)

| V (2) + Ü (1) |

### Method of assessment (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)

written examination (approx. 60 to 90 minutes)

### Allocation of places

--

### Additional information

Pursuant to Section 2 Subsection 2 Sentence 2 Verordnung über die Ausbildung und Prüfung der Staatlich geprüften Lebensmittelchemikerinnen und Lebensmittelchemiker (Regulation on the training and examination of state-certified food chemists, APOLmCh) in conjunction with No. II 2. Letter e) and No. II 1. Letter c) of Annex 1 of APOLmCh and No. 3 of Annex 3 of APOLmCh.

Referred to in LPO I (examination regulations for teaching-degree programmes)

--
Module title

Inorganic Chemistry of the Elements

Abbreviation

08-AS1-152-m01

Module coordinator

lecturer of lecture "Chemie der Hauptgruppenelemente" (Chemistry of Main-group Elements)

Module offered by

Institute of Inorganic Chemistry

ECTS

6

Method of grading

Only after succ. compl. of module(s)

numeral grade

Duration

1 semester

Module level

undergraduate

Other prerequisites

--

Contents

German contents available but not translated yet.


Intended learning outcomes

German intended learning outcomes available but not translated yet.

Der/Die Studierende kann Hauptgruppenelemente und Übergangsmetall-Elemente hinsichtlich Struktur, Reaktivität und Herstellung charakterisieren. Er/Sie ist in der Lage, die Koordination der Atome zu erkennen und zu benennen. Zudem kann er/sie das Periodensystem als grundlegendes Werkzeug in der Chemie verwenden.

Courses (type, number of weekly contact hours, language — if other than German)

V (2) + V (2)

Method of assessment (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)

a) written examination (approx. 90 to 180 minutes) or b) oral examination of one candidate each (20 to 30 minutes) or c) oral examination in groups of up to 3 candidates (approx. 15 minutes per candidate) or d) log (approx. 20 pages) or e) presentation (approx. 30 minutes)

Language of assessment: German and/or English

Allocation of places

--

Additional information

--

Referred to in LPO I (examination regulations for teaching-degree programmes)

§ 62 I Nr. 1
Module title: Organic Chemistry 1  
Abbreviation: 08-OC1-152-m01

Module coordinator: holder of the Professorship of Organic Chemistry
Module offered by: Institute of Organic Chemistry

ECTS: 5  
Method of grading: Only after succ. compl. of module(s)

Duration: 1 semester  
Module level: undergraduate
Other prerequisites: --

Contents:
German contents available but not translated yet.

Das Modul bietet einen Überblick über die elementaren Grundkenntnisse der organischen Chemie. Dazu wird die Bindungssituation am Kohlenstoff betrachtet und in die Nomenklatur einfacher und mäßig komplexer organischer Verbindungen eingeführt. Es werden Grundlagen der Stereochemie, Substitutions-, Additions- und Eliminierungsreaktionen sowie der Syntheseplanung vermittelt.

Intended learning outcomes:
German intended learning outcomes available but not translated yet.


Courses (type, number of weekly contact hours, language — if other than German):
V (3) + Ü (1)

Method of assessment (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus):

a) written examination (approx. 90 to 180 minutes) or b) oral examination of one candidate each (20 to 30 minutes) or c) oral examination in groups of up to 3 candidates (approx. 15 minutes per candidate) or d) log (approx. 20 pages) or e) presentation (approx. 30 minutes)

Language of assessment: German and/or English

Allocation of places: --

Additional information: --

Referred to in LPO I (examination regulations for teaching-degree programmes):
§ 62 I Nr. 2
**Module title**
Thermodynamics, Kinetics, Electrochemistry

**Abbreviation**
08-PC-TKE-152-m01

**Module coordinator**
Lecturer of lecture "Thermodynamik, Kinetik, Elektrochemie"

**Module offered by**
Institute of Physical and Theoretical Chemistry

<table>
<thead>
<tr>
<th>ECTS</th>
<th>Method of grading</th>
<th>Only after succ. compl. of module(s)</th>
</tr>
</thead>
<tbody>
<tr>
<td>9</td>
<td>numerical grade</td>
<td>--</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Duration</th>
<th>Module level</th>
<th>Other prerequisites</th>
</tr>
</thead>
<tbody>
<tr>
<td>1 semester</td>
<td>undergraduate</td>
<td>--</td>
</tr>
</tbody>
</table>

**Contents**

German contents available but not translated yet.


**Intended learning outcomes**

German intended learning outcomes available but not translated yet.

Die Studierenden sind in der Lage, die Hauptsätze der Thermodynamik zu erklären. Er/Sie kann thermodynamische Aspekte von Lösungen, Gasen, Mischphasen sowie elektrochemischen Reaktionen darstellen. Die Studierenden können chemische Reaktionen auf kinetischer Ebene interpretieren.

**Courses** (type, number of weekly contact hours, language — if other than German)

V (4) + Ü (2)

**Method of assessment** (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)

a) written examination (approx. 90 to 180 minutes) or b) oral examination of one candidate each (20 to 30 minutes) or c) oral examination in groups of up to 3 candidates (approx. 15 minutes per candidate) or d) log (approx. 20 pages) or e) presentation (approx. 30 minutes)

Language of assessment: German and/or English creditable for bonus

**Allocation of places**

--

**Additional information**

--

**Referred to in LPO I** (examination regulations for teaching-degree programmes)

§ 62 I Nr. 1
### Module title
Organic Chemistry 2 and analytical methods in organic chemistry

### Abbreviation
08-OC2-152-m01

### Module coordinator
holder of the Chair of Physically Organic Chemistry

### Module offered by
Institute of Organic Chemistry

### ECTS
9

### Method of grading
numerical grade

### Only after succ. compl. of module(s)
--

### Duration
1 semester

### Module level
undergraduate

### Other prerequisites
--

### Contents
German contents available but not translated yet.


### Intended learning outcomes
German intended learning outcomes available but not translated yet.


### Courses (type, number of weekly contact hours, language — if other than German)
V (3) + Ü (1) + V (2)

### Method of assessment (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)
a) written examination (approx. 90 to 180 minutes) or b) oral examination of one candidate each (20 to 30 minutes) or c) oral examination in groups of up to 3 candidates (approx. 15 minutes per candidate) or d) log (approx. 20 pages) or e) presentation (approx. 30 minutes)

Language of assessment: German and/or English

### Allocation of places
--

### Additional information
--

### Referred to in LPO I (examination regulations for teaching-degree programmes)
--
## Module title

**Practical course of Biochemistry**

### Abbreviation

08-BCP-152-m01

## Module coordinator

holder of the Chair of Biochemistry

## Module offered by

Chair of Biochemistry

## ECTS

5

## Method of grading

Only after succ. compl. of module(s)

## (not) successfully completed

08-BC1

## Duration

1 semester

## Module level

undergraduate

## Other prerequisites

--

## Contents

In this module the basics of scientific biochemical experimentation shall be practiced in practical exercises.

## Intended learning outcomes

After participating in the practical exercises the students master basic biochemical methods and are able to purposefully apply them.

## Courses (type, number of weekly contact hours, language — if other than German)

P (6)

## Method of assessment (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)

Log (approx. 30 pages)

Assessment offered: Once a year, summer semester

## Allocation of places

Students of the Bachelor's degree programme Biochemie (Biochemistry, 180 ECTS credits): no restrictions with regard to available places.

Students of the Bachelor's degree programme Chemie (Chemistry, 180 ECTS credits): no more than 6 places; places will be allocated according to the number of subject semesters, among applicants with the same number of subject semesters, places will be allocated by lot; a waiting list will be maintained and places re-allocated by lot as they become available.

## Additional information

--

## Referred to in LPO I (examination regulations for teaching-degree programmes)

--
<table>
<thead>
<tr>
<th>Module title</th>
<th>Abbreviation</th>
</tr>
</thead>
<tbody>
<tr>
<td>Organic Chemistry - advanced laboratory course for students of chemistry</td>
<td>08-OCP2-152-m01</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Module coordinator</th>
<th>Module offered by</th>
</tr>
</thead>
<tbody>
<tr>
<td>holder of the Chair of Organic Chemistry II</td>
<td>Institute of Organic Chemistry</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>ECTS</th>
<th>Method of grading</th>
<th>Only after succ. compl. of module(s)</th>
</tr>
</thead>
<tbody>
<tr>
<td>5</td>
<td>(not) successfully completed</td>
<td>08-OC2 and (08-OCP1 or OCP1-BC)</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Duration</th>
<th>Module level</th>
<th>Other prerequisites</th>
</tr>
</thead>
<tbody>
<tr>
<td>1 semester</td>
<td>undergraduate</td>
<td>--</td>
</tr>
</tbody>
</table>

**Contents**

The module deepens the experimental skills of the student by handling special hazardous substances, complex work and synthesis techniques, extensive purification methods and laborious product analyses.

**Intended learning outcomes**

The student is able to handle special hazardous substances responsibly. He/She can carry out extensive synthesis, purification techniques and product analyses. He/She can use specific literature to plan experiments.

**Courses**

<table>
<thead>
<tr>
<th>(type, number of weekly contact hours, language — if other than German)</th>
</tr>
</thead>
<tbody>
<tr>
<td>P (11)</td>
</tr>
</tbody>
</table>

**Method of assessment**

<table>
<thead>
<tr>
<th>(type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Vortestate/Nachtestate (pre and post-experiment examination talks approx. 15 minutes each, log approx. 5 to 10 pages each) and assessment of practical performance (2 to 4 random examinations)</td>
</tr>
<tr>
<td>Language of assessment: German and/or English</td>
</tr>
</tbody>
</table>

**Allocation of places**

--

**Additional information**

--

**Referred to in LPO I** (examination regulations for teaching-degree programmes)

--
<table>
<thead>
<tr>
<th>Module title</th>
<th>Abbreviation</th>
</tr>
</thead>
<tbody>
<tr>
<td>Symmetry, chemical bonding and light</td>
<td>08-PC-SBL-152-m01</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Module coordinator</th>
<th>Module offered by</th>
</tr>
</thead>
<tbody>
<tr>
<td>lecturer of lecture &quot;Symmetrie, chemische Bindung and Licht&quot;</td>
<td>Institute of Physical and Theoretical Chemistry</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>ECTS</th>
<th>Method of grading</th>
<th>Only after succ. compl. of module(s)</th>
</tr>
</thead>
<tbody>
<tr>
<td>9</td>
<td>numerical grade</td>
<td>--</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Duration</th>
<th>Module level</th>
<th>Other prerequisites</th>
</tr>
</thead>
<tbody>
<tr>
<td>2 semester</td>
<td>undergraduate</td>
<td>--</td>
</tr>
</tbody>
</table>

**Contents**

The module provides an introduction to the symmetry of molecules. It focuses on group theory, symmetry operations, point groups, character tables, and selection rules. The module deals with the chemical bond based on the qualitative MO theory and gives an introduction into the basics of computational chemistry. The module provides the opportunity to analyze the interactions between symmetry, chemical bonding and light in detail.

**Intended learning outcomes**

The student is able to analyze the symmetry of molecules. He/She can imply on the spectroscopic properties of a molecule by its symmetry.

**Courses** (type, number of weekly contact hours, language — if other than German)

V (3) + Ü (2) + V (2) + Ü (2)

**Method of assessment** (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)

a) written examination (approx. 90 to 180 minutes) or b) oral examination of one candidate each (20 to 30 minutes) or c) oral examination in groups of up to 3 candidates (approx. 15 minutes per candidate) or d) log (approx. 20 pages) or e) presentation (approx. 30 minutes)

Language of assessment: German and/or English

**Allocation of places**

--

**Additional information**

--

**Referred to in LPO I** (examination regulations for teaching-degree programmes)

--
<table>
<thead>
<tr>
<th>Module title</th>
<th>Abbreviation</th>
</tr>
</thead>
<tbody>
<tr>
<td>Analytical Chemistry (lab)</td>
<td>08-ANP-152-m01</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Module coordinator</th>
<th>Module offered by</th>
</tr>
</thead>
<tbody>
<tr>
<td>holder of the Chair of Anorganic Chemistry</td>
<td>Institute of Inorganic Chemistry</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>ECTS</th>
<th>Method of grading</th>
<th>Only after succ. compl. of module(s)</th>
</tr>
</thead>
<tbody>
<tr>
<td>6</td>
<td>(not) successfully completed</td>
<td>--</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Duration</th>
<th>Module level</th>
<th>Other prerequisites</th>
</tr>
</thead>
<tbody>
<tr>
<td>1 semester</td>
<td>undergraduate</td>
<td>--</td>
</tr>
</tbody>
</table>

**Contents**

The module provides the opportunity to apply the knowledge of the introductory lectures in a practical course. After a safety introduction the students experiment independently in the laboratory. Focuses are different analysis methods with unknown compounds.

**Intended learning outcomes**

The student is able to analyze unknown compounds using different methods. He/She is able to separate and analyze mixtures.

**Courses** (type, number of weekly contact hours, language — if other than German)

P (12) + S (1)

**Method of assessment** (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)

Vortestate/Nachtestate (pre and post-experiment examination talks approx. 15 minutes each, log approx. 5 to 10 pages each) and assessment of practical performance (2 to 4 random examinations)

Assessment offered: Once a year, summer semester

Language of assessment: German and/or English

**Allocation of places**

--

**Additional information**

--

**Referred to in LPO I** (examination regulations for teaching-degree programmes)

--
### Module title
Inorganic Chemistry 1 (lab)

### Abbreviation
08-ACP1-152-m01

### Module coordinator
holder of the Chair of Anorganic Chemistry

### Module offered by
Institute of Inorganic Chemistry

### ECTS
10

### Duration
1 semester

### Module level
undergraduate

### Other prerequisites
--

### Contents
The module provides the opportunity to apply the knowledge of the introductory lectures in a practical course. After a safety introduction the students experiment independently in the laboratory. Focuses are laboratory safety, basic laboratory techniques, synthesis of basic compounds and analysis of an unknown compound.

### Intended learning outcomes
The student is able to identify basic chemical issues and to solve them experimentally. Therefore he/she can carry out the necessary stoichiometric calculations and correctly outline the chemical processes written and verbal.

### Courses
P (12) + S (2)

### Method of assessment
a) assessment and b) Vortestate/Nachtestate (pre and post-experiment examination talks approx. 15 minutes each, log approx. 5 to 10 pages each) and assessment of practical assignments (2 to 4 random examinations)

Assessment offered: Once a year, winter semester

Language of assessment: German and/or English

### Allocation of places
--

### Additional information
--

### Referred to in LPO 1
(examination regulations for teaching-degree programmes)

--
### Module title
Solid State Chemistry, Spectroscopic Methods, Organoelement Chemistry

<table>
<thead>
<tr>
<th>Module coordinator</th>
<th>Module offered by</th>
</tr>
</thead>
<tbody>
<tr>
<td>lecturers of lecture &quot;Festkörperchemie&quot; (Solid State Chemistry) and &quot;Elementorganische Chemie&quot; (Elemental Organic Chemistry)</td>
<td>Institute of Inorganic Chemistry</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Abbreviation</th>
<th>08-AC-FSE-152-m01</th>
</tr>
</thead>
</table>

<table>
<thead>
<tr>
<th>ECTS</th>
<th>Method of grading</th>
<th>Only after succ. compl. of module(s)</th>
</tr>
</thead>
<tbody>
<tr>
<td>12</td>
<td>numerical grade</td>
<td>--</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Duration</th>
<th>Module level</th>
<th>Other prerequisites</th>
</tr>
</thead>
<tbody>
<tr>
<td>2 semester</td>
<td>undergraduate</td>
<td>--</td>
</tr>
</tbody>
</table>

### Contents
The module imparts deepened knowledge of metals, alloys, salt-like compounds and organometallics. Focuses are structure and characteristics, special substance classes, reactivity and technical processes.

### Intended learning outcomes
The student can correctly outline the structure and characteristics of metals, alloys, salt-like compounds and organometallics. He/She is able to systemize them and to characterize them with regard to their structure and reactivity. He/She can develop and describe principles of synthesis for elementorganic compounds. He/She can list and describe appropriate spectroscopical methods for the structural analysis of solid materials.

### Courses (type, number of weekly contact hours, language — if other than German)
V (2) + V (2) + V (3) + Ü (1)

### Method of assessment (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)
a) written examination (approx. 90 to 180 minutes) or b) oral examination of one candidate each (20 to 30 minutes) or c) oral examination in groups of up to 3 candidates (approx. 15 minutes per candidate) or d) log (approx. 20 pages) or e) presentation (approx. 30 minutes)

Language of assessment: German and/or English

### Allocation of places
--

### Additional information
--

### Referred to in LPO I (examination regulations for teaching-degree programmes)
--
<table>
<thead>
<tr>
<th>Module title</th>
<th>Abbreviation</th>
</tr>
</thead>
<tbody>
<tr>
<td>Organic Chemistry 3 &amp; 4</td>
<td>08-OC3+4-152-m01</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Module coordinator</th>
<th>Module offered by</th>
</tr>
</thead>
<tbody>
<tr>
<td>holder of the Professorship of Organic Chemistry</td>
<td>Institute of Organic Chemistry</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>ECTS</th>
<th>Method of grading</th>
<th>Only after succ. compl. of module(s)</th>
</tr>
</thead>
<tbody>
<tr>
<td>13</td>
<td>numerical grade</td>
<td>--</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Duration</th>
<th>Module level</th>
<th>Other prerequisites</th>
</tr>
</thead>
<tbody>
<tr>
<td>2 semester</td>
<td>undergraduate</td>
<td>--</td>
</tr>
</tbody>
</table>

### Contents

The module focuses on polar rearrangements, olefination reactions, pericyclic reactions, carbenes, nitriles and radicals. It imparts basic knowledge of stereoselective synthesis, asymmetric catalysis, organometallic chemistry and retrosynthesis. Further focuses are heterocycles, dyes, natural products, biopolymers and protecting groups.

### Intended learning outcomes

The student is able to formulate olefination reactions. He/She can develop stereoselective syntheses and asymmetric catalyses. He/She can outline organometallic reactions and analyse a molecule by retrosynthesis. He/She can name important heterocyclics and outline their reactions and synthesis. He/She is able to characterize and categorize dyes and to describe the structure and the selective synthesis of proteins. Furthermore, he/she can outline the structure of DNA, carbohydrates, fats, terpenes and steroids.

### Courses (type, number of weekly contact hours, language — if other than German)

V (2) + Ü (2) + V (2) + Ü (2) + S (1)

### Method of assessment (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)

a) written examination (approx. 90 to 180 minutes) or b) oral examination of one candidate each (20 to 30 minutes) or c) oral examination in groups of up to 3 candidates (approx. 15 minutes per candidate) or d) log (approx. 20 pages) or e) presentation (approx. 30 minutes)

Language of assessment: German and/or English

### Allocation of places

--

### Additional information

--

### Referred to in LPO I (examination regulations for teaching-degree programmes)

--
### Module title
**Principles of quantum mechanics and spectroscopy**

### Abbreviation
08-PC-QMS-152-m01

### Module coordinator
Lecturer of lecture "Grundlagen der Quantenmechanik and Spektroskopie" (Principles of Quantum Mechanics and Spectroscopy)

### Module offered by
Institute of Physical and Theoretical Chemistry

### ECTS
10

### Method of grading
Only after succ. compl. of module(s)

### Duration
1 semester

### Module level
Undergraduate

### Other prerequisites
--

### Contents
German contents available but not translated yet.


### Intended learning outcomes
German intended learning outcomes available but not translated yet.

Die Studierenden sind in der Lage, grundlegende Modelle der Quantenmechanik zu erklären und bei Molekülen anzuwenden. Er/Sie kann unterschiedliche spektroskopische Methoden darstellen. Die Studierenden können die mathematischen Grundlagen der elementaren Quantenmechanik anwenden.

### Courses (type, number of weekly contact hours, language — if other than German)

V (4) + Ü (2) + V (2)

### Method of assessment (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)

- a) written examination (approx. 90 to 180 minutes) or b) oral examination of one candidate each (20 to 30 minutes) or c) oral examination in groups of up to 3 candidates (approx. 15 minutes per candidate) or d) log (approx. 20 pages) or e) presentation (approx. 30 minutes)

Language of assessment: German and/or English creditable for bonus

### Allocation of places
--

### Additional information
--

### Referred to in LPO I (examination regulations for teaching-degree programmes)
--
<table>
<thead>
<tr>
<th><strong>Module title</strong></th>
<th><strong>Abbreviation</strong></th>
</tr>
</thead>
<tbody>
<tr>
<td>Physical Chemistry (lab)</td>
<td>08-PCP-152-m01</td>
</tr>
</tbody>
</table>

**Module coordinator**
- lecturer of lecture "Thermodynamik, Kinetik, Elektrochemie"

**Module offered by**
- Institute of Physical and Theoretical Chemistry

<table>
<thead>
<tr>
<th><strong>ECTS</strong></th>
<th><strong>Method of grading</strong></th>
<th><strong>Only after succ. compl. of module(s)</strong></th>
</tr>
</thead>
<tbody>
<tr>
<td>9</td>
<td>(not) successfully completed</td>
<td>08-PC-QMS or 08-PC-TKE</td>
</tr>
</tbody>
</table>

**Duration**
- 1 semester
- **Module level**
  - undergraduate

**Other prerequisites**

**Contents**

German contents available but not translated yet.


**Intended learning outcomes**

German intended learning outcomes available but not translated yet.


**Courses** (type, number of weekly contact hours, language — if other than German)

P (6)

**Method of assessment** (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)

Vortestate/Nachtestate (pre and post-experiment examination talks approx. 15 minutes each, log approx. 5 to 10 pages each) and assessment of practical performance (2 to 4 random examinations)

Language of assessment: German and/or English

**Allocation of places**

--

**Additional information**

--

**Referred to in LPO I** (examination regulations for teaching-degree programmes)

--
### Module Catalogue for the Subject Chemistry

**Bachelor’s with 1 major, 180 ECTS credits**

<table>
<thead>
<tr>
<th>Module title</th>
<th>Abbreviation</th>
</tr>
</thead>
<tbody>
<tr>
<td>Advanced laboratory course</td>
<td>08-VP-152-m01</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Module coordinator</th>
<th>Module offered by</th>
</tr>
</thead>
<tbody>
<tr>
<td>head of the research group offering the module</td>
<td>Faculty of Chemistry and Pharmacy</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>ECTS</th>
<th>Method of grading</th>
<th>Only after succ. compl. of module(s)</th>
</tr>
</thead>
<tbody>
<tr>
<td>5</td>
<td>(not) successfully completed</td>
<td>--</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Duration</th>
<th>Module level</th>
<th>Other prerequisites</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>undergraduate</td>
<td>--</td>
</tr>
</tbody>
</table>

**Contents**

German contents available but not translated yet.

Das Modul bietet die Möglichkeit sich mit Hilfe der für den jeweiligen Fachbereich üblichen wissenschaftlichen Arbeitstechniken und Methoden vertieft in ein Forschungsthema einzuarbeiten.

**Intended learning outcomes**

The student is able to deeply acquaint himself/herself with a specific research topic, and to process and to present the results in a written report or a talk.

**Courses** (type, number of weekly contact hours, language — if other than German)

- P (10)

**Method of assessment** (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)

- talk (approx. 15 minutes)
  - Language of assessment: German and/or English

**Allocation of places**

- --

**Additional information**

- --

**Referred to in LPO I** (examination regulations for teaching-degree programmes)

- --
### Module Title: Applied Spectroscopy 3

<table>
<thead>
<tr>
<th>Abbreviation</th>
<th>08-PS3-152-m01</th>
</tr>
</thead>
</table>

<table>
<thead>
<tr>
<th>Module Coordinator</th>
<th>Module offered by</th>
</tr>
</thead>
<tbody>
<tr>
<td>lecturer of lecture &quot;Praktische Spektroskopie 3&quot;</td>
<td>Institute of Physical and Theoretical Chemistry</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>ECTS</th>
<th>Method of Grading</th>
<th>Only after succ. compl. of module(s)</th>
</tr>
</thead>
<tbody>
<tr>
<td>5</td>
<td>numerical grade</td>
<td>--</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Duration</th>
<th>Module Level</th>
<th>Other Prerequisites</th>
</tr>
</thead>
<tbody>
<tr>
<td>1 semester</td>
<td>undergraduate</td>
<td>--</td>
</tr>
</tbody>
</table>

### Contents

German contents available but not translated yet.

Das Modul bietet die Möglichkeit, das theoretische Wissen über spektroskopische Methoden praktisch umzusetzen und die erhaltenen Messwerte bzw. Graphen zu interpretieren. Im Detail werden UV/VIS-, Fluoreszenz- und Schwingungsspektren aufgenommen sowie analysiert. Im Modul werden zudem modernste Methoden der Massenspektrometrie behandelt.

### Intended Learning Outcomes

German intended learning outcomes available but not translated yet.

Die Studierenden sind in der Lage, verschiedene Spektrometer zu bedienen und das erhaltene Spektrum zu interpretieren. Er/Sie kann eine Fehlerdiskussion durchführen.

### Courses

(type, number of weekly contact hours, language — if other than German)

V (3)

### Method of Assessment

(type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)

a) written examination (approx. 90 to 180 minutes) or b) oral examination of one candidate each (20 to 30 minutes) or c) oral examination in groups of up to 3 candidates (approx. 15 minutes per candidate) or d) log (approx. 20 pages) or e) presentation (approx. 30 minutes)

Language of assessment: German and/or English

### Allocation of Places

--

### Additional Information

--

### Referred to in LPO I

(examination regulations for teaching-degree programmes)

--
Module title          | Abbreviation
----------------------|------------------
Programming and numerical methods | 08-PKC-152-m01

Module coordinator | Module offered by
lecturer of lecture "Programmierkurs für Chemiker" | Institute of Physical and Theoretical Chemistry

ECTS | Method of grading | Only after succ. compl. of module(s)
5    | (not) successfully completed | --

Duration | Module level | Other prerequisites
1 semester | undergraduate | --

Contents
The module introduces students to the basics of a programming language and gives applications to problems related to chemistry.

Intended learning outcomes
German intended learning outcomes available but not translated yet.

Die Studierenden können einfach Grundlagen der Programmiersprache beschreiben und auf chemierelevante Probleme anwenden.

Courses (type, number of weekly contact hours, language — if other than German)
S (2) + Ü (2)

Method of assessment (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)
a) written examination (approx. 90 to 180 minutes) or b) oral examination of one candidate each (20 to 30 minutes) or c) oral examination in groups of up to 3 candidates (approx. 15 minutes per candidate) or d) log (approx. 20 pages) or e) presentation (approx. 30 minutes)
Assessment offered: Once a year, summer semester
Language of assessment: German and/or English

Allocation of places
--

Additional information
--

Referred to in LPO I (examination regulations for teaching-degree programmes)
--
<table>
<thead>
<tr>
<th>Module title</th>
<th>Abbreviation</th>
</tr>
</thead>
<tbody>
<tr>
<td>Advanced chemical practical course</td>
<td>08-OP-152-m01</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Module coordinator</th>
<th>Module offered by</th>
</tr>
</thead>
<tbody>
<tr>
<td>head of the research group offering the module</td>
<td>Faculty of Chemistry and Pharmacy</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>ECTS</th>
<th>Method of grading</th>
<th>Only after succ. compl. of module(s)</th>
</tr>
</thead>
<tbody>
<tr>
<td>5</td>
<td>(not) successfully completed</td>
<td>--</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Duration</th>
<th>Module level</th>
<th>Other prerequisites</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>undergraduate</td>
<td>--</td>
</tr>
</tbody>
</table>

**Contents**

German contents available but not translated yet.

Das Modul bietet die Möglichkeit sich mit Hilfe der für den jeweiligen Fachbereich üblichen wissenschaftlichen Arbeitstechniken und Methoden vertieft in ein Forschungsthema einzuarbeiten.

**Intended learning outcomes**

The student is able to deeply acquaint himself/herself with a specific research topic, and to process and to present the results in a written report or a talk.

**Courses** (type, number of weekly contact hours, language — if other than German)

P (10)

**Method of assessment** (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)

a) talk (approx. 15 minutes) or b) log (approx. 10 to 20 pages)

Language of assessment: German and/or English

**Allocation of places**

--

**Additional information**

--

**Referred to in LPO I** (examination regulations for teaching-degree programmes)

--
<table>
<thead>
<tr>
<th>Module title</th>
<th>Abbreviation</th>
</tr>
</thead>
<tbody>
<tr>
<td>Bachelor Thesis</td>
<td>08-BA-152-m01</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Module coordinator</th>
<th>Module offered by</th>
</tr>
</thead>
<tbody>
<tr>
<td>head of the research group offering the module</td>
<td>Faculty of Chemistry and Pharmacy</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>ECTS</th>
<th>Method of grading</th>
<th>Only after succ. compl. of module(s)</th>
</tr>
</thead>
<tbody>
<tr>
<td>10</td>
<td>numerical grade</td>
<td>--</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Duration</th>
<th>Module level</th>
<th>Other prerequisites</th>
</tr>
</thead>
<tbody>
<tr>
<td>1 semester</td>
<td>undergraduate</td>
<td>The supervisor may make the successful completion of certain modules that are relevant for the respective topic a prerequisite for the assignment of the topic.</td>
</tr>
</tbody>
</table>

**Contents**

The module enables the processing of a defined problem within a specified period by applying the scientific methods learned in the course of study.

**Intended learning outcomes**

The student has the ability to deal with a defined problem/issue using scientific methods and to document the results.

**Courses** (type, number of weekly contact hours, language — if other than German)

No courses assigned to module

**Method of assessment** (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)

Bachelor's thesis (approx. 40 pages)
Language of assessment: German and/or English

**Allocation of places**

--

**Additional information**

--

**Referred to in LPO 1** (examination regulations for teaching-degree programmes)

--
Module title | Abbreviation
---|---
Solid State Chemistry, Spectroscopic Methods (DD) | 08-AC-FS-DA-152-m01

Module coordinator

lecturer of lecture "Festkörperchemie" (Solid State Chemistry) | Module offered by
Institute of Inorganic Chemistry

ECTS | Method of grading | Only after succ. compl. of module(s)
---|---|---
5 | numerical grade | --

Duration | Module level | Other prerequisites
---|---|---
1 semester | undergraduate | --

Contents

German contents available but not translated yet.


Intended learning outcomes

German intended learning outcomes available but not translated yet.


Courses (type, number of weekly contact hours, language — if other than German)

V (2) + V (2)

Method of assessment (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)

a) written examination (approx. 90 to 180 minutes) or b) oral examination of one candidate each (20 to 30 minutes) or c) oral examination in groups of up to 3 candidates (approx. 15 minutes per candidate) or d) log (approx. 20 pages) or e) presentation (approx. 30 minutes)

Language of assessment: German and/or English

Allocation of places

--

Additional information

--

Referred to in LPO I (examination regulations for teaching-degree programmes)

--
<table>
<thead>
<tr>
<th>Module title</th>
<th>Abbreviation</th>
</tr>
</thead>
<tbody>
<tr>
<td>Organic Chemistry 3 (DD)</td>
<td>08-OC-OC3-DA-152-m01</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Holder of the Professorship of Organic Chemistry</th>
<th>Module offered by</th>
</tr>
</thead>
<tbody>
<tr>
<td>Institute of Organic Chemistry</td>
<td></td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>ECTS</th>
<th>Method of grading</th>
<th>Only after succ. compl. of module(s)</th>
</tr>
</thead>
<tbody>
<tr>
<td>6</td>
<td>numerical grade</td>
<td>--</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Duration</th>
<th>Module level</th>
<th>Other prerequisites</th>
</tr>
</thead>
<tbody>
<tr>
<td>1 semester</td>
<td>undergraduate</td>
<td>--</td>
</tr>
</tbody>
</table>

**Contents**

The module focuses on polar rearrangements, olefination reactions, pericyclic reactions, carbenes, nitriles and radicals. It imparts basic knowledge of stereoselective synthesis, asymmetric catalysis, organometallic chemistry and retrosynthesis.

**Intended learning outcomes**

German intended learning outcomes available but not translated yet.


**Courses** (type, number of weekly contact hours, language — if other than German)

V (2) + Ü (2)

**Method of assessment** (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)

a) written examination (approx. 90 to 180 minutes) or b) oral examination of one candidate each (20 to 30 minutes) or c) oral examination in groups of up to 3 candidates (approx. 15 minutes per candidate) or d) log (approx. 20 pages) or e) presentation (approx. 30 minutes)

Language of assessment: German and/or English

**Allocation of places**

--

**Additional information**

--

**Referred to in LPO I** (examination regulations for teaching-degree programmes)

--
### Module title
Symmetry, chemical bonding and light (DD)

### Abbreviation
08-PC-SBL-DA-152-m01

### Module coordinator
Lecturer of lecture "Symmetrie, chemische Bindung and Licht"

### Module offered by
Institute of Physical and Theoretical Chemistry

### ECTS
6

### Method of grading
Numerical grade

### Only after succ. compl. of module(s)
--

### Duration
1 semester

### Module level
Undergraduate

### Other prerequisites
--

## Contents
The module provides an introduction to the symmetry of molecules. It focuses on group theory, symmetry operations, point groups, character tables, and selection rules. The module deals with the chemical bond based on the qualitative MO theory and gives an introduction into the basics of computational chemistry.

## Intended learning outcomes
The student is able to analyze the symmetry of molecules. He/She can imply on the spectroscopic properties of a molecule by its symmetry.

## Courses
(type, number of weekly contact hours, language — if other than German)

<table>
<thead>
<tr>
<th>Type</th>
<th>Number of Weekly Contact Hours</th>
<th>Language</th>
</tr>
</thead>
<tbody>
<tr>
<td>V</td>
<td>(3)</td>
<td></td>
</tr>
<tr>
<td>Ü</td>
<td>(2)</td>
<td></td>
</tr>
</tbody>
</table>

## Method of assessment
(type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)

- a) written examination (approx. 90 to 180 minutes) or
- b) oral examination of one candidate each (20 to 30 minutes) or
- c) oral examination in groups of up to 3 candidates (approx. 15 minutes per candidate) or
- d) log (approx. 20 pages) or
- e) presentation (approx. 30 minutes)

Language of assessment: German and/or English

## Allocation of places
--

## Additional information
--

## Referred to in LPO I
(examination regulations for teaching-degree programmes)

--
### Module Catalogue for the Subject Chemistry

**Bachelor’s with 1 major, 180 ECTS credits**

<table>
<thead>
<tr>
<th>Module title</th>
<th>Abbreviation</th>
</tr>
</thead>
<tbody>
<tr>
<td>Qualifications - Partner University 1</td>
<td>08-VPUB1-152-m01</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Module coordinator</th>
<th>Module offered by</th>
</tr>
</thead>
<tbody>
<tr>
<td>programme coordinator of the exchange programme</td>
<td>Faculty of Chemistry and Pharmacy</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>ECTS</th>
<th>Method of grading</th>
<th>Other prerequisites</th>
</tr>
</thead>
<tbody>
<tr>
<td>25</td>
<td>Only after succ. compl. of module(s)</td>
<td>Please consult with course advisory service in advance.</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Duration</th>
<th>Module level</th>
<th>Other prerequisites</th>
</tr>
</thead>
<tbody>
<tr>
<td>2 semester</td>
<td>undergraduate</td>
<td>Please consult with course advisory service in advance.</td>
</tr>
</tbody>
</table>

### Contents

The topics covered in this module correspond to the syllabus of the foreign partner university.

### Intended learning outcomes

German intended learning outcomes available but not translated yet.

Die Studierenden erwerben Kompetenzen entsprechend den besuchten Veranstaltungen an der Partneruniversität.

### Courses

No courses assigned to module

Course(s) as specified by partner university abroad

### Method of assessment

Assessments as specified by partner university abroad

Language of assessment: German and/or language spoken at partner university abroad

### Allocation of places

--

### Additional information

--

### Referred to in LPO I (examination regulations for teaching-degree programmes)

--
<table>
<thead>
<tr>
<th><strong>Module title</strong></th>
<th><strong>Abbreviation</strong></th>
</tr>
</thead>
<tbody>
<tr>
<td>Qualifications - Partner University 2</td>
<td>08-VPUB2-152-m01</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th><strong>Module coordinator</strong></th>
<th><strong>Module offered by</strong></th>
</tr>
</thead>
<tbody>
<tr>
<td>programme coordinator of the exchange programme</td>
<td>Faculty of Chemistry and Pharmacy</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th><strong>ECTS</strong></th>
<th><strong>Method of grading</strong></th>
<th><strong>Only after succ. compl. of module(s)</strong></th>
</tr>
</thead>
<tbody>
<tr>
<td>25</td>
<td>numerical grade</td>
<td>--</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th><strong>Duration</strong></th>
<th><strong>Module level</strong></th>
<th><strong>Other prerequisites</strong></th>
</tr>
</thead>
<tbody>
<tr>
<td>2 semester</td>
<td>undergraduate</td>
<td>Please consult with course advisory service in advance.</td>
</tr>
</tbody>
</table>

**Contents**

The topics covered in this module correspond to the syllabus of the foreign partner university.

**Intended learning outcomes**

German intended learning outcomes available but not translated yet.

Die Studierenden erwerben Kompetenzen entsprechend den besuchten Veranstaltungen an der Partneruniversität.

**Courses** (type, number of weekly contact hours, language — if other than German)

No courses assigned to module

**Method of assessment** (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)

Assessments as specified by partner university abroad

Language of assessment: German and/or language spoken at partner university abroad

**Allocation of places**

--

**Additional information**

--

**Referred to in LPO I** (examination regulations for teaching-degree programmes)

--
### Module title
Mathematics for students in Chemistry and Biochemistry

### Abbreviation
10-M-MCH-172-m01

### Module coordinator
Dean of Studies Mathematik (Mathematics)

### Module offered by
Institute of Mathematics

### ECTS
5

### Method of grading
numerical grade

### Only after succ. compl. of module(s)
--

### Duration
1 semester

### Module level
undergraduate

### Other prerequisites
--

### Contents
Functional relations, differentiation and integration of functions in one variable, curve sketching, differentiation and integration of functions in several variables, curve integrals, matrix calculus, power series.

### Intended learning outcomes
The student is able to recognise and phrase questions from natural sciences as mathematical problems, apply basic mathematical methods to them and interpret the results.

### Courses
(type, number of weekly contact hours, language — if other than German)
V (3) + Ü (2)

### Method of assessment
(type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)
written examination (approx. 90 to 120 minutes) and written exercises (approx. 25)

### Allocation of places
--

### Additional information
--

### Referred to in LPO I
(examination regulations for teaching-degree programmes)
--
<table>
<thead>
<tr>
<th>Module title</th>
<th>Abbreviation</th>
</tr>
</thead>
<tbody>
<tr>
<td>Inorganic Chemistry 2 (lab)</td>
<td>08-ACP2-172-m01</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Module coordinator</th>
<th>Module offered by</th>
</tr>
</thead>
<tbody>
<tr>
<td>holder of the Chair of Anorganic Chemistry</td>
<td>Institute of Inorganic Chemistry</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>ECTS</th>
<th>Method of grading</th>
<th>Only after succ. compl. of module(s)</th>
</tr>
</thead>
<tbody>
<tr>
<td>5</td>
<td>(not) successfully completed</td>
<td>(08-OCP1 or 08-OCP1-BC) and 08-AS1</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Duration</th>
<th>Module level</th>
<th>Other prerequisites</th>
</tr>
</thead>
<tbody>
<tr>
<td>1 semester</td>
<td>undergraduate</td>
<td>--</td>
</tr>
</tbody>
</table>

### Contents

The module provides the opportunity to plan and carry out complex syntheses after an individual research. Focuses are the handling of organometallic compounds, their synthesis and the work in inert atmospheres. Spectroscopical methods are used for the precise determination of the products.

### Intended learning outcomes

The student is able to experimentally solve complex issues after an individual research. He/She can describe the technical backgrounds and explain them written and verbal using technical language. He/She can independently plan and carry out the synthesis of a chemical compound. Therefore he/she can apply advanced laboratory techniques.

### Courses

(type, number of weekly contact hours, language — if other than German)

P (12)

### Method of assessment

(type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)

Vortestate/Nachtestate (pre and post-experiment examination talks approx. 15 minutes each, log approx. 5 to 10 pages each) and assessment of practical performance (2 to 4 random examinations)

Language of assessment: German and/or English

### Allocation of places

--

### Additional information

--

**Referred to in LPO I** (examination regulations for teaching-degree programmes)

--
Module title | Abbreviation
---|---
Organic Chemistry - lab 1 | 08-OCP1-172-m01

Module coordinator | Module offered by
holder of the Chair of Organic Chemistry II | Institute of Organic Chemistry

ECTS | Method of grading | Only after succ. compl. of module(s)
8 | (not) successfully completed | 08-OC1 and (08-ACP1 or 08-ANP)

Duration | Module level | Other prerequisites
1 semester | undergraduate | --

Contents
German contents available but not translated yet.


Intended learning outcomes
German intended learning outcomes available but not translated yet.


Courses (type, number of weekly contact hours, language — if other than German)
P (14)

Method of assessment (type, scope, language — if other than German, examination offered — if not every semester, information on whether module can be chosen to earn a bonus)

Vortestate/Nachtestate (pre and post-experiment examination talks approx. 15 minutes each, log approx. 5 to 10 pages each) and assessment of practical performance (2 to 4 random examinations)

Language of assessment: German and/or English

Allocation of places
--

Additional information
--

Referred to in LPO I (examination regulations for teaching-degree programmes)
--